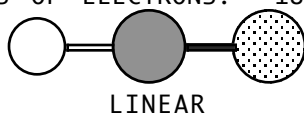
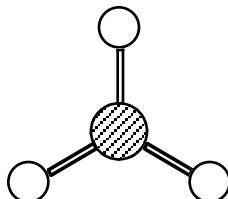


VALENCE SHELL ELECTRON PAIR REPULSION

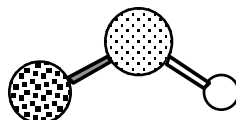
TWO PAIRS OF ELECTRONS: 180° ANGLES



THREE PAIRS OF ELECTRONS: 120° ANGLES

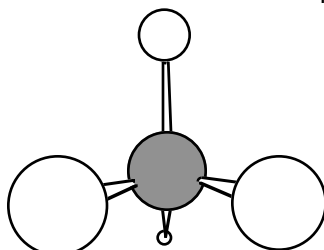


TRIGONAL PLANAR

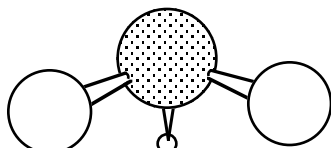


BENT (V SHAPED)

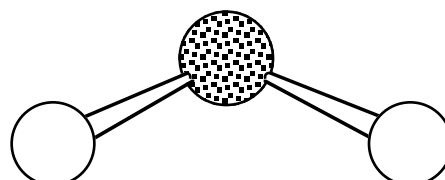
FOUR PAIRS OF ELECTRONS: 109.5° ANGLES



TETRAHEDRAL

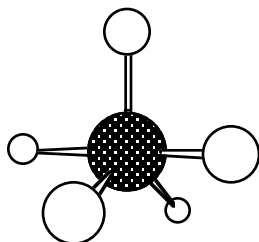


TRIGONAL PYRAMIDAL

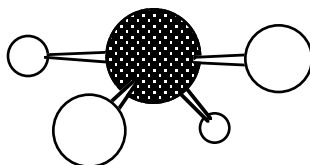


BENT (V SHAPED)

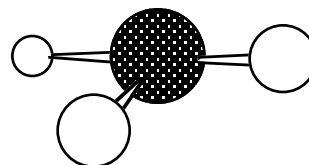
FIVE PAIRS OF ELECTRONS: 120° AND 90° ANGLES



TRIGONAL BIPYRAMIDAL

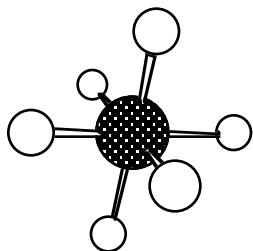


SEESAW

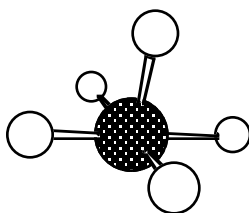


T SHAPED

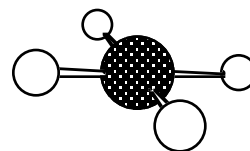
SIX PAIRS OF ELECTRONS: 90° ANGLES



OCTAHEDRAL



SQUARE PYRAMIDAL



SQUARE PLANAR